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Crystal Data: Monoclinic. Point Group: 2/m. Crystals, to 0.6 mm, slightly elongated along [010], showing poorly developed $\{001\}$, $\{101\}$, $\{100\}$, $\{201\}$, $\{110\}$; commonly equidimensional granular.

Physical Properties: Cleavage: Perfect on $\{001\}$, $\{100\}$; distinct on $\{110\}$. Fracture: Subconchoidal to uneven. Tenacity: Very brittle. Hardness = 2.5-3 D(meas.) = 2.88 D(calc.) = 2.891 Soluble in H₂O, with separation of Fe(OH)₃ from a weakly acid solution.

Optical Properties: Transparent. Color: Very pale purplish pink; colorless in transmitted light. Streak: White to pale yellow. Luster: Vitreous to nearly adamantine. Optical Class: Biaxial (-). Orientation: Z = b; $X \wedge c = 6^{\circ}$. Dispersion: r > v, strong, horizontal. $\alpha = 1.593(2)$ $\beta = 1.684(2)$ $\gamma = 1.698(1)$ $2V(\text{meas.}) = 30.5^{\circ}$

Cell Data: Space Group: C2/m. a = 8.152(5) b = 5.153(4) c = 7.877(5) $\beta = 94.90(7)^{\circ}$ Z = 2

X-ray Powder Pattern: Jerome, Arizona, USA. 2.97 (100), 7.85 (90), 3.87 (80), 2.394 (80), 3.73 (75), 2.842 (75), 3.49 (60)

Chemistry:

	(1)	(2)
SO_3	55.84	55.78
Al_2O_3	0.11	
Fe_2O_3	27.82	27.81
FeO	0.07	
MgO	0.08	
Na_2O	0.62	
K_2O	15.46	16.41
Total	[100.00]	100.00

(1) Jerome, Arizona, USA; recalculated to 100% from an original total of 99.94% after deduction of $\rm H_2O^-$ 0.14% and $\rm SiO_2$ + insoluble 0.26%; corresponds to $\rm (K_{0.93}Na_{0.06})_{\Sigma=0.99}$ (Fe_{1.00}Mg_{0.01})_{\Sigma=1.01}(SO₄)_{2.00}. (2) KFe(SO₄)₂.

Occurrence: In the fire zone of a pyritic orebody, formed under fumarolic conditions.

Association: Voltaite, sulfur, jarosite.

Distribution: From the open pit at the United Verde mine, Jerome, Yavapai Co., Arizona, USA.

Name: For the Yavapai Indian tribe who inhabit the region around Jerome, Arizona, USA.

Type Material: n.d.

References: (1) Hutton, C.O. (1959) Yavapaiite, an anhydrous potassium, ferric sulphate from Jerome, Arizona. Amer. Mineral., 44, 1105–1114. (2) Graeber, E.J. and A. Rosenzweig (1971) The crystal structures of yavapaiite, KFe(SO₄)₂, and goldichite, KFe(SO₄)₂•4H₂O. Amer. Mineral., 56, 1917–1933. (3) Anthony, J.W., W.J. McLean, and R.B. Laughon (1972) The crystal structure of yavapaiite: a discussion. Amer. Mineral., 57, 1546–1549.