Crystal Data: Monoclinic. *Point Group*: 2/*m*. As blocky to short prismatic crystals to 0.2 mm. *Twinning*: On {001} confirmed by X-ray analysis.

Physical Properties: Cleavage: Good on $\{001\}$.Fracture: Splintery.Tenacity: Brittle.Hardness = ~ 4 D(meas.) = 2.76(4)D(calc.) = 2.772

Optical Properties: Transparent. *Color*: Brownish orange. *Streak*: White. *Luster*: Vitreous. *Optical Class*: Biaxial (–). $\alpha = 1.629(4)$ $\beta = 1.658(4)$ $\gamma = 1.677(4)$ $2V(calc.) = 76.8^{\circ}$ *Pleochroism*: Weak; Y = orange-pink, Z = pale orange, X = very pale gray. *Absorption*: Y > Z > X.

Cell Data: Space Group: P2/a. a = 14.975(5) b = 7.1645(14) c = 9.928(2) $\beta = 110.65(3)^{\circ}$ Z = 2

X-ray Powder Pattern: Tom's quarry, Kapunda, South Australia. 9.339 (100), 2.839 (35), 4.923 (20), 3.562 (20), 3.518 (20), 3.453 (20), 2.965 (20)

Chemistry:	(1)
Na ₂ O	0.65
CaO	4.74
MgO	6.47
MnO	5.02
FeO	[9.85]
Fe ₂ O ₃	[20.18]
Al_2O_3	0.06
P_2O_5	34.41
H ₂ O	[19.46]
Total	100.84

(1) Tom's quarry, Kapunda, South Australia; average of 17 electron microprobe analyses supplemented by IR spectroscopy, H₂O calculated from structure, Fe₂O₃ and FeO calculated from crystal-chemical constraints; corresponds to $(Ca_{0.70}Na_{0.17}Mn^{2+}_{0.16})_{\Sigma=1.03}Fe^{2+}_{1.00}(Mg_{1.33}Mn^{2+}_{0.43}Fe^{3+}_{0.24})_{\Sigma=2.00}(Fe^{3+}_{1.99}Al_{0.01})_{\Sigma=2.00}(PO_4)_{4.01}(OH)_{2.10}H_2O_{7.88}.$

Mineral Group: Whiteite-jahnsite group.

Occurrence: A secondary mineral in low-grade phosphorites derived by leaching of weakly phosphatic limestones or low-grade primary phosphorites.

Association: Jahnsite-(NaFeMg), goethite, fluorapatite.

Distribution: From Tom's quarry, Koonunga Hill area, 10 km E of Kapunda, South Australia.

Name: For a jahnsite group mineral with dominant Ca in the X site and Fe^{2+} and Mg in the M1 and M2 sites.

Type Material: South Australian Museum, Adelaide, South Australia, Australia (G34045).

References: (1) Elliot, P. (2016) Jahnsite-(CaFeMg), a new mineral from Tom's quarry, South Australia: description and crystal structure. Eur. J. Mineral., 28(6), 991-996. (2) (2017) Amer. Mineral., 102, 1962-1963 (abs. ref. 1).