

**Jahnsite-(NaMnMg)****(Na, Ca)Mn<sup>2+</sup>(Mg, Fe<sup>3+</sup>)<sub>2</sub>Fe<sup>3+</sup><sub>2</sub>(PO<sub>4</sub>)<sub>4</sub>(OH)<sub>2</sub>·8H<sub>2</sub>O**

**Crystal Data:** Monoclinic. *Point Group:* 2/m. As prisms to 0.5 mm, elongate along [100], exhibiting {100}, {001}, and {011}. *Twinning:* By reflection on {001}.

**Physical Properties:** *Cleavage:* Perfect on {001}. *Fracture:* Irregular, stepped splintery. *Tenacity:* Brittle. Hardness =~4 D(meas.) = 2.68(1)-2.69(3) D(calc.) = 2.684-2.738 Slowly soluble in dilute HCl.

**Optical Properties:** Translucent to transparent. *Color:* Light orange to orange-yellow. *Streak:* Very pale yellow. *Luster:* Vitreous. *Optical Class:* Biaxial (-).  $\alpha = 1.642$   $\beta = 1.675$   $\gamma = 1.677$  2V(meas.) = 27(2) $^\circ$  *Orientation:* Z = b; X  $\wedge$  c = 51 $^\circ$  in obtuse  $\beta$ . *Pleochroism:* X = colorless, Y = Z = orange-yellow. *Absorption:* Y  $\approx$  Z > X. *Dispersion:* Very strong,  $r < v$ .

**Cell Data:** *Space Group:* P2/a.  $a = 15.1045(15)$   $b = 7.1629(2)$   $c = 9.8949(7)$   $\beta = 110.640(7)^\circ$   $Z = 2$

**X-ray Powder Pattern:** Sapucaia pegmatite, Conselheiro Pena district, Minas Gerais, Brazil. 9.29 (100), 2.834 (91), 2.601 (33), 1.944 (33), 3.546 (32), 4.91 (30), 5.02 (27)

Chemistry:	(1)	(2)
Na <sub>2</sub> O	2.16	2.35
CaO	1.73	1.53
MgO	7.64	6.15
MnO	[8.27]	[10.72]
Mn <sub>2</sub> O <sub>3</sub>	0	[3.94]
Fe <sub>2</sub> O <sub>3</sub>	23.83	20.77
Al <sub>2</sub> O <sub>3</sub>	1.31	0.13
P <sub>2</sub> O <sub>5</sub>	35.23	34.02
H <sub>2</sub> O	20.31	19.45
Total	100.48	99.06

(1) Sapucaia pegmatite, Minas Gerais, Brazil; average electron microprobe analysis, H<sub>2</sub>O and Mn apportioned from structure analysis; corresponds to (Na<sub>0.56</sub>Ca<sub>0.25</sub>Mn<sup>2+</sup><sub>0.09</sub>)<sub>Σ=0.90</sub>(Mn<sup>2+</sup><sub>0.85</sub>Fe<sup>3+</sup><sub>0.15</sub>)<sub>Σ=1.00</sub>(Mg<sub>1.53</sub>Fe<sup>3+</sup><sub>0.47</sub>)<sub>Σ=2.00</sub>(Fe<sup>3+</sup><sub>1.79</sub>Al<sub>0.21</sub>)<sub>Σ=2.00</sub>(PO<sub>4</sub>)<sub>4</sub>(OH)<sub>1.83</sub>(H<sub>2</sub>O)<sub>8.17</sub>. (2) White Rock No. 2 quarry, South Australia, Australia; average electron microprobe analysis, H<sub>2</sub>O and Mn apportioned from structure analysis; corresponding to (Na<sub>0.63</sub>Ca<sub>0.23</sub>Mn<sup>2+</sup><sub>0.14</sub>)<sub>Σ=1.00</sub>(Mn<sup>2+</sup><sub>0.68</sub>Mn<sup>3+</sup><sub>0.26</sub>Fe<sup>3+</sup><sub>0.05</sub>Mg<sub>0.01</sub>)<sub>Σ=1.00</sub>(Mg<sub>1.26</sub>Mn<sup>2+</sup><sub>0.43</sub>Mn<sup>3+</sup><sub>0.16</sub>Fe<sup>3+</sup><sub>0.15</sub>)<sub>Σ=2.00</sub>(Fe<sup>3+</sup><sub>1.97</sub>Al<sub>0.02</sub>)<sub>Σ=1.99</sub>(PO<sub>4</sub>)<sub>4</sub>(OH)<sub>1.98</sub>(H<sub>2</sub>O)<sub>8.02</sub>.

**Mineral Group:** Jahnsite group, jahnsite subgroup; Fe<sup>3+</sup> > Al in the M(3) structural site.

**Occurrence:** A low temperature, secondary mineral formed by alteration of primary phosphates in zoned granitic pegmatite.

**Association:** Frondelite, albite, meurigite-K, phosphosiderite, strengite (Sapucaia); phosphosiderite, ushkovite, strunzite, bermanite (White Rock); jahnsite-(NaFeMg) (Tip Top).

**Distribution:** In the Sapucaia pegmatite, 12 km north northeast of Galiléia, Conselheiro Pena district, Minas Gerais, Brazil and the White Rock No. 2 quarry, Bimbawrie Conservation Park, ~22 km north of Olary, South Australia, Australia. At the Tip Top pegmatite, South Dakota, USA.

**Name:** Root name, *Jahnsite*, indicates a member of the group with M(3) = Fe<sup>3+</sup>, the suffix indicates sequentially the dominant atom in the X, M(1), and M(2) structural positions.

**Type Material:** Natural History Museum of Los Angeles County, Los Angeles, California, USA (66701) and the South Australian Museum, Adelaide, South Australia (G34298).

**References:** (1) Kampf, A.R., P. Elliott, B.P. Nash, L. Chiappino, and S. Varvello (2018) Jahnsite-(NaMnMg), a new Jahnsite-group mineral from the Sapucaia Mine, Brazil and the White Rock No. 2 quarry, Australia. Can. Mineral., 56(6), 871-882. (2) (2021) Amer. Mineral., 106, 1362 (abs. ref. 1).