

**Crystal Data:** Hexagonal. *Point Group:* 6. Prismatic crystals to 1.5 mm, display {100} and {001}, in radiating groups. As acicular crystals to 0.2 mm, in nearly parallel or sheaf-like clusters, also as open chaotic aggregates to 4 mm, that resemble matted wool. In cauliflower-like clusters, to 1 cm [xenoliths in volcanic glass].

**Physical Properties:** *Cleavage:* Distinct on {100}. *Fracture:* Uneven or stepped.

*Tenacity:* Brittle. *Hardness* = 2-2.5 D(meas.) = 1.82(3) D(calc.) = 1.86

Slowly dissolves in HCl with very weak effervescence.

**Optical Properties:** Transparent. *Color:* Colorless; snow-white aggregates. *Streak:* White.

*Luster:* Vitreous; silky aggregates.

*Optical Class:* Uniaxial (-).  $\omega = 1.494(2)$   $\varepsilon = 1.476(2)$

**Cell Data:** *Space Group:*  $P6_3$ ,  $a = 11.1178(2)$   $c = 10.5381(2)$   $Z = 2$

**X-ray Powder Pattern:** Graulay quarry, Rhineland-Palatinate, Germany.

9.62 (100), 3.823 (64), 5.551 (50), 2.742 (38), 4.616 (37), 2.528 (37), 2.180 (35)

Chemistry:	(1)	(2)
CaO	27.15	26.93
Al <sub>2</sub> O <sub>3</sub>	2.33	
SiO <sub>2</sub>	7.04	9.62
SO <sub>3</sub>	[12.91]	12.82
SO <sub>2</sub>	[6.40]	10.25
CO <sub>2</sub>	2.71	
N <sub>2</sub> O <sub>5</sub>	0.42	
H <sub>2</sub> O	39.22	40.38
Total	98.18	100.00
SO <sub>3</sub> (meas.)	20.91	

(1) Graulay quarry, Rhineland-Palatinate, Germany; average of 8 electron microprobe analyses, complex anions confirmed by IR spectroscopy, SO<sub>2</sub> and SO<sub>3</sub> recalculated from SO<sub>3</sub>(meas.) based on structural analysis, H<sub>2</sub>O, CO<sub>2</sub>, and N<sub>2</sub>O<sub>5</sub> by gas chromatography and CHN analysis; corresponding to  $\text{Ca}_3(\text{Si}_{0.73}\text{Al}_{0.28})_{\Sigma=1.01}(\text{OH})_{5.71}(\text{SO}_4)_{1.00}(\text{SO}_3)_{0.62}(\text{CO}_3)_{0.38}(\text{NO}_3)_{0.05} \cdot 10.63\text{H}_2\text{O}$ .

(2)  $\text{Ca}_3\text{Si}(\text{OH})_6(\text{SO}_4)(\text{SO}_3) \cdot 11\text{H}_2\text{O}$ .

**Polymorphism & Series:** Forms a series with thaumasite.

**Occurrence:** In miarolitic cavities in alkaline basalt or also in xenoliths in volcanic glass.

**Association:** Diopside, nepheline, fluorapatite, magnetite, phillipsite-K, chabazite-Ca, gypsum, gismondine.

**Distribution:** From Graulay quarry near Hillesheim, western Eifel Mountains, Rhineland-Palatinate, Germany.

**Name:** Honors German mineral collector Klaus Hielscher (b. 1957) from Steinbach, Hessen, a specialist in the mineralogy of the Zeilberg basalt quarry in Franconia, Bavaria, Germany.

**Type Material:** A.E. Fersman Mineralogical Museum, Russian Academy of Sciences, Moscow, Russia (#4093/1).

**References:** (1) Pekov, I.V., N.V. Chukanov, S.N. Britvin, Y.K. Kabalov, J. Göttlicher, V.O. Yapaskurt, A.E. Zadov, S.V. Krivovichev, W. Schüller, and B. Ternes (2012) The sulfite anion in ettringite-group minerals: a new mineral species hielscherite,  $\text{Ca}_3\text{Si}(\text{OH})_6(\text{SO}_4)(\text{SO}_3) \cdot 11\text{H}_2\text{O}$ , and the thaumasite-hielscherite solid-solution series. *Mineral. Mag.*, 76(5), 1133-1152. (2) (2015) *Amer. Mineral.*, 100, 1324-1325 (abs. ref. 1).